

## Answers - Review Problems for Exam 2

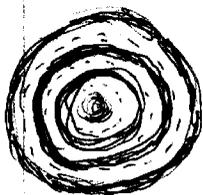
1a.  $u_{rms}$  depends on temp and molar mass,  $\propto \sqrt{\frac{1}{M}}$   
 $u_{rms} = \sqrt{\frac{3RT}{M}}$   
 At the same temp, lighter gas molecules travel faster. So He would have a higher root-mean-square speed than  $CO_2$ .

b. false - in any sample, molecules are traveling at a wide range of speeds. For He, the average speed will be higher, but there will always be some very slow and some very fast molecules in both samples.

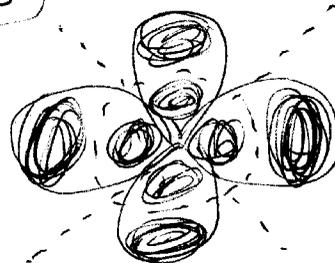
2. 2d doesn't exist:  $n=2$  and  $l=2$  but  $l$  can't be equal to or greater than  $n$ .  
 7s and 5d are ok.

2f doesn't exist:  $n=2$ ,  $l=3$  - same reason as above

3. 4s  $n-l-1 = 4-0-1 = 3$  radial nodes



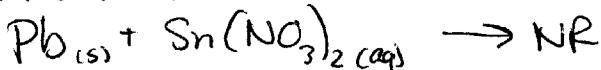
4d  
 $n-l-1 =$   
 $4-2-1 = 1$  radial node  
 $l = 2$  nodal planes



4a) Al is more active than H, so this rxn will happen.



b) Tin is more active than lead, so no rxn for



c) This rxn will happen since Sn is more active than Pb.



5. a. not allowed. If  $l=0$ ,  $m_l$  can only = 0  
 $m_l = 0, \pm 1, \pm 2, \dots, \pm l$

b. allowed. This would be a 3d orbital.  
 $l < n$ ,  
 $m_l$  is in the correct range (-2, -1, 0, 1, 2)  
 $m_s$  is fine