Chemistry 30B Experiment 15

| Lab Instructor: | Name: |  |
|-----------------|-------|--|
|                 |       |  |
|                 |       |  |

## Part 1 – Solubility of Vitamins

| Vitamin | Soluble in CH <sub>2</sub> Cl <sub>2</sub> ? | Soluble in water?    | Water-soluble or |
|---------|--|----------------------|------------------|
|         | Record observations.                         | Record observations. | fat-soluble?     |
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## Part 2 – Standardization of an Iodine Solution by Titration with Vitamin C

| Quantity                              | Result (include units) |
|---------------------------------------|------------------------|
| Mass of Vitamin C (from label)        |                        |
|                                       |                        |
| Initial Buret Reading (1)             |                        |
| Final Buret Reading (1)               |                        |
| Initial Buret Reading (2) (If needed) |                        |
| Final Buret Reading (2)               |                        |
| Initial Buret Reading (3) (If needed) |                        |
| Final Buret Reading (3)               |                        |
| Total volume of iodine solution used  |                        |

| Calculations (show calculation setups)                     | Result (include units) |
|--|------------------------|
| Mass of Vitamin C that reacts with 1 mL of iodine solution |                        |
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|  | 1                      |

## Part 3 – Analysis of Vitamin C in Fruit Juices and Fruit Drinks

| Data (include units)      | Sample 1 | Sample 2 |
|---------------------------|----------|----------|
| Type of juice, drink, or  |          |          |
| vegetable                 |          |          |
| Amount of juice, drink or |          |          |
| vegetable used            |          |          |
| Initial buret reading     |          |          |
|                           |          |          |
| Final buret reading       |          |          |
|                           |          |          |
| Volume of iodine solution |          |          |
| used                      |          |          |

| Calculations (show calculation setups)        | Result (include units) |
|---|------------------------|
| Number of milligrams of vitamin C in Sample 1 |                        |
|   |                        |
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| Number of milligrams of vitamin C in Sample 2 |                        |
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## Part 4 – Heat Destruction of Vitamin C

| Data (include units)             | Heated 10 minutes | Heated 30 minutes |
|----------------------------------|-------------------|-------------------|
| Type of juice, drink, or         |                   |                   |
| Vegetable (same for both trials) |                   |                   |
| Amount of juice, drink or        |                   |                   |
| vegetable used                   |                   |                   |
| Initial buret reading            |                   |                   |
|                                  |                   |                   |
| Final buret reading              |                   |                   |
|                                  |                   |                   |
| Volume of iodine solution        |                   |                   |
| used                             |                   |                   |

| Calculations (show calculation setups)                           | Result (include units) |
|--|------------------------|
| Number of milligrams of vitamin C in sample after heating for 10 |                        |
| minutes:   |                        |
|  |                        |
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| Numb<br>minute | er of milligrams of vitamin C <b>destroyed</b> after heating for 10 es:  |
|----------------|--|
| Numb           | er of milligrams of vitamin C in sample after heating for 30 es:   |
| Numb<br>minute | er of milligrams of vitamin C <b>destroyed</b> after heating for 30 es:  |
| Ques           | tions  |
| 1.             | Which of the vitamins you tested were water-soluble? Which were fat-soluble?   |
| 2.             | Which vitamins would be excreted daily?  |
| 3.             | Which of the juices that you analyzed had the highest vitamin C content?   |
| 4.             | If the daily requirement of vitamin C is 75 mg, how many milliliters (or grams) of each sample would you need to consume to get 75 mg? Show your calculations. |

| 5. | How does heating affect the vitamin C content of a fruit juice?  |
|----|--|
| 6. | If vitamin C tablets are stored in a warm, humid bathroom cabinet, what might happen to the vitamin C content after a while? |
| 7. | If you wanted to keep most of the vitamin C content of your vegetables, how would you prepare them for dinner?               |
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