Chem 30B

Ch17: Carboxylic Acids and Their Derivatives

Carboxylic Acid Compounds

Volatile acids have strong, sharp odors.

Ester Compounds

Volatile esters have pleasant, fruity odors.

(Ethyl butanoate)

pineapple



Pear, strawberry, jasmine scent

pine

apple

Polyamide Compounds

Amides are generally odorless.

$$\begin{array}{c}
n \text{ HOOC} - (\text{CH}_2)_4 - \text{COOH} \\
\text{Adipic acid} \\
+ \\
n \text{ Hexamethylenediamine}
\end{array}$$

$$\begin{array}{c}
280^\circ \\
-\text{H}_2\text{O}
\end{array}$$

$$\begin{array}{c}
\text{C} - (\text{CH}_2)_4 - \text{C} - \text{NH} - (\text{CH}_2)_6 - \text{NH} \\
\text{Nylon 6,6, a polyamide} \\
\text{(repeating unit)}
\end{array}$$

© 2013 Pearson Education, Inc.

Peptides, proteins

Carboxylic Acid Derivative Drugs

$$\begin{array}{c} O \\ H_2N - \begin{array}{c} O \\ - \\ C - O - CH_2CH_3 \end{array} \\ \begin{array}{c} CH_3 \\ - \\ NHCCH_2N(CH_2CH_3)_2 \end{array} \\ \\ CH_3 \\ CH_3 \\ \\$$

© 2013 Pearson Education, Inc.

Common Names for Some Carboxylic Acids

TABLE 17.1 Physical Properties of Some Carboxylic Acids

STRUCTURE	COMMON NAME	MELTING POINT (°C)	BOILING POINT (°C)
Carboxylic Acids			
НСООН	Formic	8	101
CH₃COOH	Acetic	17	118
CH ₃ CH ₂ COOH	Propionic	-22	141
CH ₃ CH ₂ CH ₂ COOH	Butyric	-4	163
CH ₃ CH ₂ CH ₂ COOH	Valeric	-34	185
CH ₃ (CH ₂) ₁₆ COOH	Stearic	70	383
Dicarboxylic Acids			
НООССООН	Oxalic	190	Decomposes
HOOCCH₂COOH	Malonic	135	Decomposes
HOOCCH ₂ CH ₂ COOH	Succinic	188	Decomposes
HOOCCH ₂ CH ₂ CH ₂ COOH	Glutaric	98	Decomposes
Unsaturated Acids			
H ₂ C=CHCOOH	Acrylic	13	141
CH₃CH=CHCOOH	Crotonic	72	185
Aromatic Acids			
СООН	Benzoic	122	249
СООН	Salicylic	159	Decomposes