## **Handout: Families of Organic Compounds**

	<b>FAMILY</b>	FUNCT GROUP	<b>EXAMPLE</b>	<u>SUFFIX</u>
Hydrocarbons	Alkane	-C-C- H-C-H   l or H   (only single bonds)	H <sub>3</sub> C-CH <sub>3</sub> ethane	-ane
	Alkene	C=C(	H <sub>2</sub> C=CH <sub>2</sub> ethylene	-ene
	Alkyne	—C≡C—	HC≡CH ethyne (acetylene)	-yne
	Aromatic	, c, c, c,	H H,C,C,C,H H,C,C,C,H H,C,C,C,H	

Compounds with single bonds only, and C bonded to electronegative atom (-X, -O,-N)	Alkyl halide	-C-X (X= F, Cl, Br, I)	H <sub>3</sub> C-Cl methyl chloride	
	Alcohol	—С-О-Н 	H <sub>3</sub> C-OH methanol	-ol
	Ether	-C-O-C-	H <sub>3</sub> C-O-CH <sub>3</sub> dimethyl ether	
	Amine (basic)	_C_N_	H <sub>3</sub> C-NH <sub>2</sub> methylamine	-amine

<sup>\*</sup>We will also learn about thiols (-C-SH), sulfides (-C-S-C-), and disulfides (-C-S-S-C-).

Carbonyl compounds (-C=O)	Aldehyde	_С-С—Н	О Н <sub>3</sub> С-С-ОН	-al
(-0-0)	Ketone	0	ethanal (acetaldehyde)	-one
	Retolle	-c-c-c-	H <sub>3</sub> C-C-CH <sub>3</sub>	-one
		1 1	(acetone)	
	Carboxylic Acid (acidic)	_с-с-о-н	O H <sub>3</sub> C-C-O-CH <sub>3</sub>	-ic acid
		l	acetic acid	
	Anhydride	-ç-c-o-c-ç-	O O H <sub>3</sub> C-C-O-C-CH <sub>3</sub>	
		I I	acetic anhydride	
	Ester	-C-C-O-C-	O H <sub>3</sub> C-C-O-CH <sub>3</sub>	-ate
			methyl acetate	
	Amide		O H <sub>3</sub> C-C-NH <sub>2</sub>	-amide
			acetamide	