Chemistry 1A Experiment 15

Lab Instructor:	Name:

SAMPLE DATA TABLES

PART 1: HEAT OF FUSION OF ICE

	Trial 1	Trial 2	Trial 3
1. Mass of empty calorimeter			
2. Mass of calorimeter plus water			
3. Mass of water in calorimeter			
4. Initial temperature of water in calorimeter (± 0.1°C)			
5. Temperature after ice has melted (± 0.1°C)			
6. Mass of calorimeter and contents after ice has melted			
7. Mass of ice			

PART 2: HEAT OF SOLUTION OF TWO SOLUTES

NH ₄ Cl	Trial 1	Trial 2	Trial 3
8. Mass of empty calorimeter			
9. Mass of calorimeter and water			
10. Mass of water in calorimeter			
11. Mass of solid used			
12. Initial temperature of water in calorimeter (± 0.1°C)			
13. Final temperature of solution in calorimeter (± 0.1°C)			
Na ₂ CO ₃	Trial 1	Trial 2	Trial 3
8. Mass of empty calorimeter			
9. Mass of calorimeter and water			
10. Mass of water in calorimeter			
11. Mass of solid used			
12. Initial temperature of water in calorimeter (± 0.1°C)			
13. Final temperature of solution in calorimeter (± 0.1°C)			

ABOVE IS A SAMPLE DATA TABLE: these go in your lab book. Do not turn the above in as your data table. The above is a sample.

CALCULATIONS

Show all calculation setups, including units in your lab book. Keep the calculations for each trial separate.

SAMPLE RESULTS TABLE

Fill this out with the results of your calculations. THIS IS A SAMPLE

This this out with the results of your calculations. This is it shift Les							
Part 1 Heat of fusion of ice in J/g	Trial 1	Trial 2	Trial 3				
ΔH_{ice}							
Average ΔH _{ice}							
$\%$ error ΔH_{ice}							
% difference ΔH_{ice}							
Part 2: Heat of solution, NH ₄ Cl	Trial 1	Trial 2	Trial 3				
$\Delta H_{solution}$							
Average $\Delta H_{solution}$							
$\%$ difference $\Delta H_{solution}$							
Part 2: Heat of solution, Na ₂ CO ₃	Trial 1	Trial 2	Trial 3				
$\Delta H_{solution}$							
Average $\Delta H_{solution}$							
% difference $\Delta H_{solution}$							

QUESTIONS:

Found on the pre-lab sheet

TURNING IT IN:

Attach these completed sheets to the lab pages which contain your observations, data tables, and calculations. Print out a cover sheet.